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Comparative analysis of the electronic structures of mono- and bi-atomic chains of IV, III–V and II–VI group elements calculated using the DFT LCAO and LACW methods

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Using the first principle non-relativistic linear combination of atomic orbitals (LCAO) and relativistic linearized augmented cylindrical wave (LACW) methods, the band structure of the covalent and partially ionic $A^N B^{8-N}$ single atom width chain is calculated. Both the LCAO and LACW methods show that the chains of C, Si, Ge, Sn, and Pb are metallic. However, there is a great difference between the relativistic and non-relativistic band structures. The π bands crossing the Fermi level are orbitally doubly degenerate in the non-relativistic model. The relativistic LACW calculations demonstrate that the spin and orbital motion of electrons are coupled, thereby splitting the π bands. The spin–orbit gaps are equal to 1.5 meV, 28 meV, 0.22 eV, 0.45 eV, and 4 eV for the C, Si, Ge, Sn, and Pb chains, respectively. The mass–velocity corrections result in a lowering of all the valence band levels. In the carbon and silicon chains, the corrections are possibly negligible (2–5 and 10–30 meV, respectively), while in the Ge, Sn, and Pb chains the low-energy shifts are equal to 0.6, 2.2, and 3.7 eV, respectively, due to these effects. The Darwin corrections are several times smaller in comparison to the mass–velocity contributions. The transition from the covalent chains to the partially ionic ones is accompanied by a drastic change in the band structure. The C chain with all bond lengths equal has a metal type electronic structure while the BN chain is an insulator with an energy gap equal to 6–8 eV. The differences between the covalent and partially ionic chains are explained by the presence of the antisymmetric components of the electron potential in the latter case. The transition from the BN chain to the AlP, GaAs, and InSb ones is accompanied by a gradual decrease in the gaps; for example, the AlP chain is a semiconductor. According to the LCAO calculations, the GaAs chain is a semiconductor, but it is a metal according to the relativistic LACW method. The InSb chain possesses a metal type band structure, but the spin–orbit interaction splits the π states, forming the two π^+ and π^- sub-bands, and noticeably complicates the band structure and density of states in the vicinity of the Fermi level. In the case of compounds from the same horizontal row in the periodic table, the transition from the $A^{III}B^V$ chains to the $A^{II}B^{VI}$ ones is accompanied by a sharp increase in the band gap. The calculations indicate the metallic nature of the InSe chain, but the CdTe one is an insulator. Among the atomic $A^N B^{8-N}$ chains, there are compounds with different electrical properties: from metals to semiconductors and insulators.

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1. Introduction

Within the current trends of miniaturization of electronic devices, the problems of interconnects between nanodevices attract growing interest in the stability, band structure, and conductivity of the nanowires. The thinnest possible nanowire is a single-atom width chain.^{1–11} The structures and properties of atomic wires are completely different from those of large

dimension wires, and they are the ideal subjects for investigating quantum effects. It still remains a challenge to fabricate stable atomic chains; however, there is great progress in this field now. First, gold chains generated inside a transmission electron microscope have been reported,^{1–3,11–14} and stable gold wires of single-atom width up to ten-atom length were formed upon the tensile deformation of the nanometer-sized Au contacts. The structures of the Au chains were determined, and the stress–strain curves and electrical properties, including ballistic transport, were experimentally measured at the atomic scale.¹⁴ Platinum wires of single-atom width exhibiting straight shapes were produced by the retraction of a Pt nanotip from contact with a Pt plate at room temperature inside

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a transmission electron microscope, and a relationship between the atomic chain length and conductance was demonstrated.^{14–16} Binary atomic chains were generated *via* Au_{1–x}Ag_x alloy nanorod thinning during mechanical stretching.¹⁷ Ligand-supported linear chains of transition metal atoms with well-defined organic ends are known too.¹⁸ A linear chain with 44 C atoms in compound (Tr–C₄₄–Tr with bulky terminal groups Tr = tris(3,5-di-*t*-butylphenyl)methyl),¹⁹ as well as similar short chains connecting two graphene species have been detected.^{20,21} Carbon nanotubes can be utilized as sheaths to stabilize monoatomic chains, which are unstable alone, and can be applied in many fields. For example, entrapped La atoms arrange linearly with a typical chain length equal to 10 nm inside a nanotube of a suitable diameter.²² The incorporation of molten iodine into single-walled carbon nanotubes with a diameter of 1 nm generates iodine chains longer than 10 nm.²³ A carbon chain with a length of 20 nm containing more than 100 atoms enclosed in a shell of multi-walled carbon nanotubes was found as well.²⁴ Inorganic structures are also known to self-assemble into one-dimensional structures on a bare Si(001) surface. This includes elemental adsorbates such as indium,^{25–28} lead,²⁹ bismuth,³⁰ and rare earth metals³¹ that form nanowires.

Monoatomic chains are well-known as text-book examples studied using the simple tight-binding or free electron models. Now, there are many first-principles calculations of the electronic structure of such systems. The band structures and magnetic properties of all 3d transition metal infinite periodic linear and dimerized linear chains were calculated using the *ab initio* pseudopotential plane-wave method (PPW) taking spin-orbit coupling into account.⁹ Employing the linearized augmented plane-wave method considering non-perturbative spin-orbit coupling, virtually the same properties of all 4d transition metal chains were studied.¹⁰ The application of the projector augmented-wave method to the problems of transition metal chains³ show that both Pt and Au stand out as the most likely transition elements to form chains.^{11,14,15,32} In the case of some transition metal atoms, a variation of the total energy as a function of the lattice constant for the different magnetic states shows that the energy of infinite linear structures has a minimum for ferromagnetic and antiferromagnetic freestanding chain structures, opening new perspectives in controlling the spin-dependent ballistic conductance.^{9–11,33,34}

In contrast to carbon and transition metal monoatomic wires,^{4–8,28–31} which have been thoroughly investigated, not much is known about atomic chains made from group IV (Si, Ge, Sn), group III–V (BN, AlP, GaAs, InSb, *etc.*), and group II–VI (ZnSe, CdTe, *etc.*) elements, which form well-known bulk semiconductor crystals. To the best of our knowledge, there is only one study about the structure and electron properties of these linear chains,^{35,36} where it was shown, using first-principles PPW calculations, that all chains are stable, the binding energies of which being close to bulk cohesive energies. The band structure calculations of a dozen A^NB^{8–N} monoatomic chains have also been performed earlier,^{35,36} but only in terms of a non-relativistic approach. However, because of the rotational symmetry of the linear atomic A^NB^{8–N} chains, relativistic corrections to the band structure are expected to be of great

importance.^{30,37} Recently, the quantum conductance of short linear monoatomic Si and Ge chains was calculated,^{38–40} and an all-silicon linear chain NMR quantum computer was suggested.

In this study, using the standard linear combination of localized orbitals (LCAO) method, we have calculated the bond lengths, formation energies, and band structures of covalent and partially ionic A^NB^{8–N} chains. As relativistic calculations are not yet implemented into the LCAO based program, we are currently calculating the band structures taking into account the spin-orbit, Darwin, and mass-velocity corrections using the original linearized augmented cylindrical wave (LACW) method.^{41–45} The LACW method is just a reformulation of the linearized augmented plane wave (LAPW) formalism for cylindrical multiatomic systems. The relativistic version of the LACW method can be readily obtained based on the relativistic APW techniques.⁴⁶

2. Linear combination of atomic orbitals

First principles LCAO calculations of chains have been performed using the formalism of atom-centered Gaussian-type functions (GTFs). According to the LCAO approach, the crystalline orbitals $\varphi_{\mathbf{k}i}(\mathbf{r})$ of an N -electron system are expanded as linear combinations of a set of m Bloch functions built from the local atom-centered GTFs $\chi_{g\mathbf{j}}(\mathbf{r} - \mathbf{R}_j)$:

$$\varphi_{\mathbf{k}i}(\mathbf{r}) = N \sum_{j=1}^m a_{ij}(\mathbf{k}) \sum_{\mathbf{g}} \chi_{g\mathbf{j}}(\mathbf{r}) \exp(i\mathbf{k}\mathbf{g}), \quad (1)$$

$$\chi_{g\mathbf{j}}(\mathbf{r} - \mathbf{R}_j) = \sum_{\mu} c_{\mu} G(\alpha_{\mu}; \mathbf{r} - \mathbf{R}_j - \mathbf{g}), \quad (2)$$

where \mathbf{k} is the wave vector of the irreducible representation of the group of crystal translations $\{\mathbf{g}\}$, \mathbf{R}_j denotes the coordinates of the nuclei in the zero cell, in which the atomic orbital $\chi_{g\mathbf{j}}(\mathbf{r})$ is centered; and G , c_{μ} , and α_{μ} are the normalized GTFs, their coefficients and exponents, respectively, forming the basis set (BS) of an individual atom.

The DFT-LCAO method realized for various Hamiltonians, as implemented in the CRYSTAL-14 code⁴⁷ and used in our calculations, differs when using the exchange-correlation functionals applied in Kohn–Sham equations:

$$\hat{v}_{\text{xc}} = \frac{\partial E_{\text{xc}}[\rho(\mathbf{r}); \mathbf{k}]}{\partial \rho(\mathbf{r})}, \quad (3)$$

where E_{xc} is the energy functional while $\rho(\mathbf{r})$ is the electron density function expressed *via* crystalline orbitals $\varphi_{\mathbf{k}i}(\mathbf{r})$. Our calculations on atomic chains were performed using the hybrid Hartree–Fock/Kohn–Sham (HF/KS) exchange–correlation functional HSE by Heyd–Scuseria–Ernzerhof^{48,49} combining an exact HF non-local exchange and KS exchange operator within the generalized gradient approximation as implemented in the CRYSTAL-14 code.⁴⁷ The advantage of the hybrid HSE-LCAO calculation scheme is that it makes the results of the band structure calculations more plausible. An all-valence TZVP (Triple-Zeta Valence with Polarization) basis set^{47,50} was used for

all atoms in the current study. The threshold parameters of the CRYSTAL code for the evaluation of different types of bielectronic integrals (overlap and penetration tolerances for Coulomb integrals, ITOL1 and ITOL2, overlap tolerance for exchange integrals ITOL3, and pseudo-overlap tolerances for exchange integral series, ITOL4 and ITOL5) have been set to 8, 8, 8, 8, and 16, respectively. They provide truncation of the corresponding integrals when the overlap between the arbitrary atomic orbitals does not exceed 10^{-ITOL_n} .

To achieve balanced summation over the direct and reciprocal lattices, reciprocal space integration was performed by sampling the Brillouin zone with the $10 \times 1 \times 1$ Pack–Monkhorst k -mesh⁵¹ which results in six evenly distributed k -points within the irreducible segment of the Brillouin zone. Calculations of atomic chains with a fixed geometry are considered to be converged only when the total energy differs by less than 10^{-7} a.u. in two successive cycles of the self-consistency procedure. The next step of the computational procedure realized using CRYSTAL is total geometry optimization which includes the optimization of the electronic structure for each intermediate and final atomic configurations.⁴⁷ Further increase in both the threshold parameters and k -meshes results in more expensive calculations yielding only a negligible gain in the total energy ($\sim 10^{-7}$ a.u.) which can be ignored for qualitative conclusions. Table 1 shows the equilibrium bond lengths d and formation energies E^{form} for nanofilaments of single-atom width calculated using this method. The LCAO band structures are presented in Sec. 4 together with similar data obtained using the LACW technique.

3. Linearized augmented cylindrical wave method

Relativistic LACW band structure calculations were performed for the geometries optimized using the LCAO method. We started from the two-component Hamiltonian written in terms of Rydberg atomic units:^{46,52–54}

$$H = -\Delta + V + \frac{1}{c^2} \boldsymbol{\sigma} \cdot [(\nabla V) \times \mathbf{p}] + \frac{1}{2c^2} \nabla^2 V - \frac{1}{c^2} p^4. \quad (4)$$

Table 1 Equilibrium bond lengths d and formation energies E^{form} of linear chains as calculated by means of the LCAO-DFT approach. Negative formation energy corresponds to an energy gain

Type of chain	d , Å	E^{form} , eV
C	1.27	−6.74
Si	2.18	−2.77
Ge	2.35	−2.03
Sn	2.52	−2.89
Pb	2.75	−1.97
BN	1.30	−7.88
AlP	2.05	−4.70
GaAs	2.16	−2.21
InSb	2.70	−0.98
ZnSe	2.18	−3.10
CdTe	2.61	−3.13

Here, c is the speed of light, p is the momentum operator, and $\boldsymbol{\sigma}$ is the Pauli matrix. The first two terms correspond to the non-relativistic Hamiltonian operator $H_0 = -\Delta + V$. Methods for solving the non-relativistic Schrödinger equation, using the basis of the linear augmented cylindrical waves $\Psi_{\text{PMN}}^k(\mathbf{r})$, are described in previous work.^{41–43} The results of this calculation are the non-relativistic energies $E_n^0(k)$ of different bands n at different points k in the Brillouin zone and the spin independent wave functions $\Psi_{n,k}^0(\mathbf{r})$ represented as the linear combination of the basis functions:

$$\Psi_{n,k}^0(\mathbf{r}) = \sum_{\text{PMN}} d_{\text{PMN}}^{kn} \Psi_{\text{PMN}}^k(\mathbf{r}). \quad (5)$$

In eqn (4), the third term is the operator of the spin–orbit interaction $H_{\text{S-O}}$, which may split degenerated levels and cause the mixing of levels, thus altering the band picture in comparison with non-relativistic results. The fourth and fifth terms (the Darwin H_{Dar} and mass–velocity $H_{\text{m-v}}$ corrections, respectively) are invariant under the operations of the single group and therefore do not split levels, but these terms may mix levels of the same single-group symmetry and shift non-relativistic levels appreciably.

As the non-relativistic part of the Hamiltonian makes the major contribution to the energy, it is possible to use the following procedure.⁴⁶ First, we find the eigenfunctions $\Psi_{n,k}^0(\mathbf{r})$ and eigenvalues $E_n^0(k)$ of the non-relativistic Hamiltonian.^{8,41–45} Then, we double the basis to include the spin $\Psi_{n,k}^0(\mathbf{r}, \chi) = \Psi_{n,k}^0(\mathbf{r})\chi$ where $\chi = \alpha$ or β are the pure spin functions. The calculation of the elements of the $H_{\text{S-O}}$, H_{Dar} , and $H_{\text{m-v}}$ matrices using the spinor basis $\Psi_{n,k}^0(\mathbf{r})\chi$ (see Appendix section)

$$\begin{aligned} \langle \Psi_{n_2,k}^0(\mathbf{r})\chi_2 | H | \Psi_{n_1,k}^0(\mathbf{r})\chi_1 \rangle &= E_{n_1,k}^0 \delta_{n_2,n_1} \delta_{\chi_2,\chi_1} \\ &+ \delta_{\chi_2,\chi_1} \langle \Psi_{n_2,k}^0(\mathbf{r}) | H_{\text{Dar}} | \Psi_{n_1,k}^0(\mathbf{r}) \rangle + \delta_{\chi_2,\chi_1} \langle \Psi_{n_2,k}^0(\mathbf{r}) | H_{\text{m-v}} | \Psi_{n_1,k}^0(\mathbf{r}) \rangle \\ &+ \langle \Psi_{n_2,k}^0(\mathbf{r})\chi_2 | H_{\text{S-O}} | \Psi_{n_1,k}^0(\mathbf{r})\chi_1 \rangle \end{aligned} \quad (6)$$

and diagonalization of this matrix remain to be done.

In the LACW method, we use the muffin-tin (MT) potentials and local density functional approximations for electronic potentials constructed to be spherically symmetrical in the regions of the MT spheres and constant in the inter-spherical area. The atoms of the nanofilaments are surrounded by an impenetrable cylinder-shaped potential barrier, because there is a vacuum region outside the wires. The radius a of the barrier is chosen so that the region confined within the barrier accommodates a significant portion of the electron density of the system. Here, based on previous LACW calculations on nanotubes and carbynes,^{8,41–47} for the fourth group chains, the barrier radii $a(\text{Å}^{\text{IV}})$ were set equal to the arithmetic mean value of the atomic covalent and van der Waals radii. For the partially ionic chains, the corresponding radii $a(\text{Å}^{\text{N}}\text{B}^{8-\text{N}})$ were taken as being the same as for the covalent analogs: $a(\text{Å}^{\text{N}}\text{B}^{8-\text{N}}) = a(\text{Å}^{\text{IV}})$.

A good convergence of the relativistic band structures is obtained using 45–50 basis functions for the monoatomic group IV chains and using 70–80 functions in the case of chains with two atoms per unit cell.

4. Results of the calculations

Fig. 1 shows the band structures of the group IV covalent chains calculated using the non-relativistic LCAO and relativistic LACW methods. The points $k = 0$ and $k = \pi/d$ correspond to the Brillouin zone Γ center and K boundary, respectively. The generic band structures of these atomic wires are composed of the σ (s) bonding, σ^* (p_z) antibonding, and one π band which is bonding at $k < \pi/(2d)$ and antibonding between $k = \pi/(2d)$ and $k = \pi/d$. Similar to the non-relativistic PPW method,^{35,36} both the LCAO and LACW techniques predict that the chains of C, Si, Ge, Sn, and Pb are metallic (Table 2). However, there is a great difference between the relativistic and non-relativistic band structures. Because of the cylindrical symmetry of the chains, the π bands crossing the Fermi level are predicted to be orbitally doubly degenerate in the non-relativistic model; the twofold orbitally degenerate π bands correspond to the semi-classical clockwise and anticlockwise rotational motion of electrons around the symmetry axis. In the absence of spin-orbit interaction with the two possible directions of the spin, the π bands would be fourfold degenerate. The LACW calculations demonstrate that the spin and orbital motion of the electrons are strongly coupled in the atomic chains, thereby splitting the fourfold degeneracy seen in Fig. 1 in the case of the relativistic LACW results. However, each π^+ and π^- sub-band of the relativistic model still has twofold degeneracy; the Kramers theorem on time-reversal symmetry alongside the inversion symmetry of chains preserves the spin degeneracy, the spin polarization direction between two degenerate bands being opposite to each other. The spin-orbit splitting energy Δ_{s-o} varies between about 1.5 meV and 0.45 eV for the C and Sn chains, respectively. In the case of the chain of the heaviest element Pb, Δ_{s-o} is already equal to 4 eV. There is no spin-orbit splitting of the orbitally nondegenerate σ bands.

The LACW data show that the mass-velocity corrections result in the lowering of all the valence band levels. In the C and Si chains, the shifts are equal to 2–5 and 10–30 meV for different k points and are possibly negligible, but the mass-velocity shifts increase up to 0.6, 2.2, and 3.7 eV in the cases of the Ge, Sn, and Pb atomic wires. The Darwin corrections are several times smaller in comparison to the mass-velocity contributions, and can somewhat shift the energy levels to the high or low-energy region.

In the Fermi energy region, there are important differences between the band structures of the carbon and other group IV chains; the result was also observed with the non-relativistic PPW method.^{35,36} In the case of the C chain, the π bands cross the Fermi level at $k = \pi/2d$ giving rise to Peierls' distortion, whereas the Si, Ge, Sn, and Pb chains behave differently; in addition to the π bands, the σ^* band dips below the Fermi level near the zone edge preventing Peierls' dimerization. The LCAO optimization of the atomic positions reveals that bond length alternation is possible only in the case of the C chain, resulting in the formation of the so called polyynic carbyne semi-conducting chain $(-C\equiv C-)_\infty$, with the calculated $C\equiv C$ and $C-C$ bonds being equal to 1.23 and 1.32 Å, respectively. Fig. 2 shows

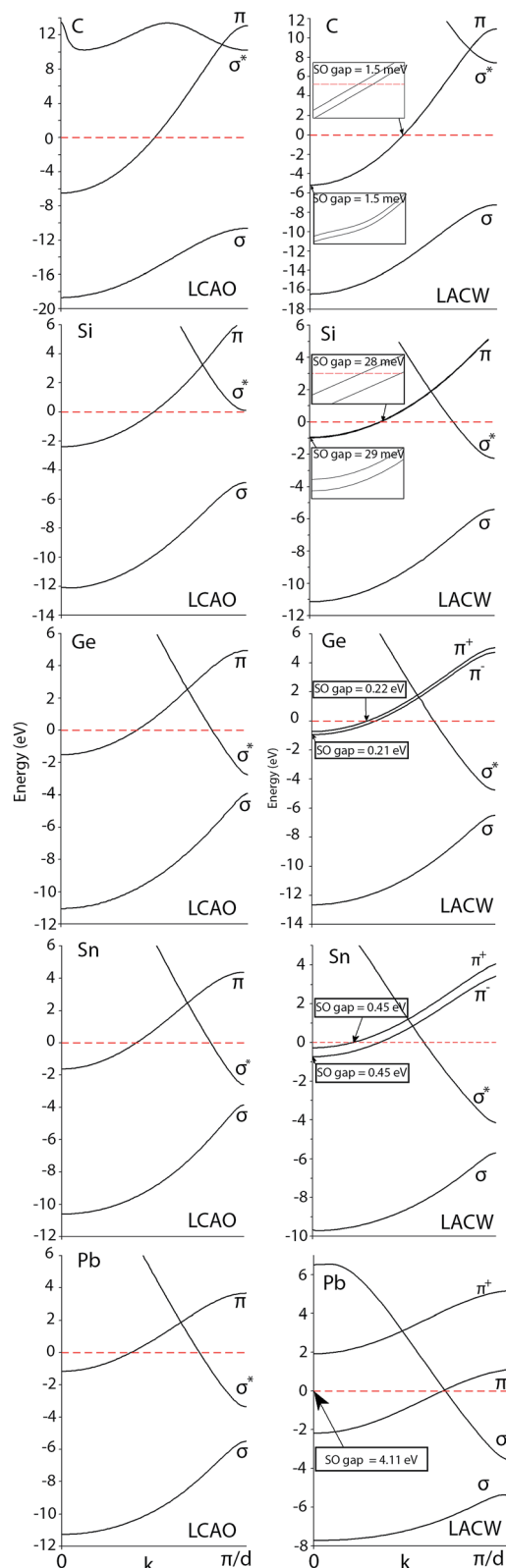


Fig. 1 Band structures of the linear group IV chains calculated using the non-relativistic LCAO and relativistic LACW methods. The zero of energy is set at the Fermi level.

Table 2 Electronic characteristics of the linear chains in the Fermi energy region calculated using the non-relativistic PPW and LCAO and relativistic LACW methods. The energy gaps between the valence and conduction bands (E_g) and spin-orbit gaps (Δ_{SO}) are given in eV. The symmetries π , π^+ , π^- , and σ of the dispersion curves crossing the Fermi level of the metallic wires or corresponding to the top of the valence and bottom of the conduction bands are shown too

	C	Si	Ge	Sn	Pb
PPW ^{35,36}	$E_g = 0$; π	$E_g = 0$; π , σ^*	$E_g = 0$; π , σ^*	$E_g = 0$; π , σ^*	$E_g = 0$; π , σ^*
LCAO	$E_g = 0$; π	$E_g = 0$; π	$E_g = 0$; π , σ^*	$E_g = 0$; π , σ^*	$E_g = 0$; π , σ^*
LACW	$E_g = 0$, π ; $\Delta_{SO} = 0.0015$	$E_g = 0$; π^+ , π^- , σ^* ; $\Delta_{SO} = 0.028$	$E_g = 0$; π^+ , π^- , σ^* ; $\Delta_{SO} = 0.22$	$E_g = 0$; π^+ , π^- , σ^* ; $\Delta_{SO} = 0.45$	$E_g = 0$; π^+ , π^- , σ^* ; $\Delta_{SO} = 4.1$
	BN	AlP	GaAs	InSb	
PPW ^{35,36}	$E_g = 4$; $\pi_K \rightarrow \pi_K^+$	$E_g = 1.1, 1.6$; $\pi_K \rightarrow \sigma_1^*$	$E_g = 0$; π , σ^*	$E_g = 0$; π , σ^*	
LCAO	$E_g = 6$; $\pi_K \rightarrow \pi_K^+$	$E_g = 3.5$; $\pi_K \rightarrow \pi_K^+$	$E_g = 0.5$; $\pi_K \rightarrow \sigma_1^*$	$E_g = 0$; π , σ^*	
LACW	$E_g = 8$; $\pi_K \rightarrow \pi_K^+$; $\Delta_{SO} = 0.0015$	$E_g = 2.2$; $\pi_K \rightarrow \sigma_1^*$; $\Delta_{SO} = 0.35$	$E_g = 0$; π^+ , σ^* ; $\Delta_{SO} = 0.26$	$E_g = 0$; π^+ , σ^* ; $\Delta_{SO} = 0.51$	
	ZnSe	CdTe			
PPW ^{35,36}	$E_g = 0.7, 0.8$; $\pi_K \rightarrow \sigma_1^*$	$E_g \approx 0$; $\pi_K \rightarrow \sigma_1^*$			
LCAO	$E_g = 3.5$; $\pi_K \rightarrow \sigma_1^*$	$E_g = 2$; $\pi_K \rightarrow \sigma_1^*$			
LACW	$E_g = 6.5$; $\pi_K \rightarrow \sigma_1^+$; $\Delta_{SO} = 0.3$	$E_g = 0.8$; $\pi_K \rightarrow \sigma_1^+$; $\Delta_{SO} = 0.6$			

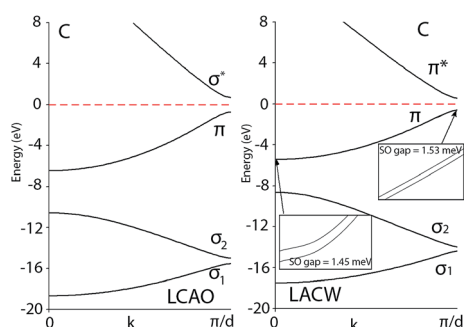


Fig. 2 Non-relativistic LCAO and relativistic LACW band structures of polynic carbyne.

the LCAO and LACW band structures for this semiconducting phase.

Fig. 3 and Table 2 show that the transition from the covalent chains to the partially ionic ones is accompanied by a sharp change in the band structure. For example, the carbon chain with all bond lengths equal to 1.27 Å has a metal type electronic structure with a zero gap located at the point $k = \pi/2d$, but the boron nitride chain with almost the same bond lengths of 1.30 Å is an insulator with an optical gap corresponding to the transition between the occupied π and vacant π^* states at the edge of the Brillouin zone and equal to 4, 6, and 8 eV in the PPW, LCAO, and LACW calculations, respectively. Qualitatively, the differences between the electronic bands of the covalent and partially ionic chains are explained by the presence of the antisymmetric components of the electron potential in the transition from the carbon chain to the boron nitride one. This is also the reason for a similar splitting of the σ bands in the BN chain. The transition from the atomic BN chain to the AlP, GaAs, and InSb chains is accompanied by a decrease in the ionic

character of the chemical bond leading to a gradual decrease in the π - π^* gap and shift of the σ_1^* bands in the region of lower energies. These effects lead to the fact that the AlP chain is a semiconductor with an indirect band gap equal to 1.1–1.6 (PPW), 2.2 eV (LACW) or 3.5 eV (LCAO).

According to the LCAO calculations, the chain GaAs is a semiconductor with a band gap equal to 0.5 eV; the gap is due to a transition between the π state on the edge of the Brillouin zone and the σ state in the center. It is a metal according to both the PPW and relativistic LACW methods; the Fermi level crosses the π^+ zone, a formation which is associated with spin-orbit splitting. All methods predict that the InSb chain has a metal type band structure due to the intersection of the π and σ^* bands; the spin-orbit interaction splits the π states, forms the two π^+ and π^- bands, and noticeably complicates the band structure and density of states in the Fermi level region. The chemical bonding in the compounds $A^{III}B^{VI}$ is much more ionic than in $A^{III}B^V$; therefore, the antisymmetric component of the electron potential in the $A^{III}B^V$ chains is stronger, and, in the case of compounds from the same horizontal row in the periodic table, the transition from the $A^{III}B^V$ chains to the $A^{III}B^{VI}$ ones is accompanied by a sharp increase in the band gap. For example, Fig. 3 and 4 show that, in the GaAs chain, the gap is absent or equal to about 0.5 eV, based on different data, but it is large in the ZnSe chain. All calculations indicate the metal type of the band structure of the InSe chain, but CdTe is an insulator with a band gap equal to 2 and 0.8 eV in the non-relativistic LCAO calculations and in the LACW calculations with due account of relativistic corrections, respectively.

It is interesting to underscore that the presence of the ionic component of the electronic potential dramatically changes the band structure in the transition from the covalent to partially ionic chains, although the ionicity almost does not affect the energy of spin-orbit splitting, which is equal to about 2 meV in

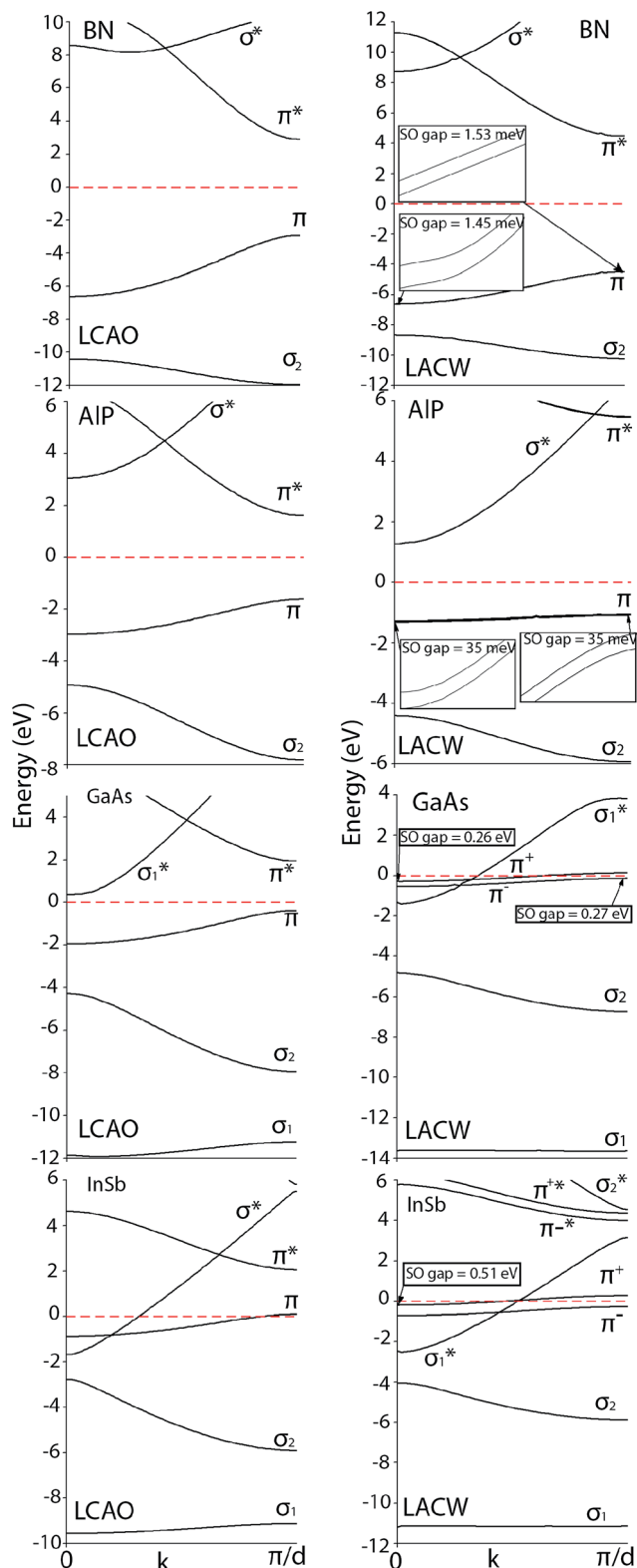


Fig. 3 Non-relativistic LCAO and relativistic LACW band structures of the $A^{III}B^V$ chains.

the chains of the elements of the second period, of about 30 meV in the chains of the elements of the third period, of 0.3 eV in the chains of the elements of the fourth period, and 0.5 eV in

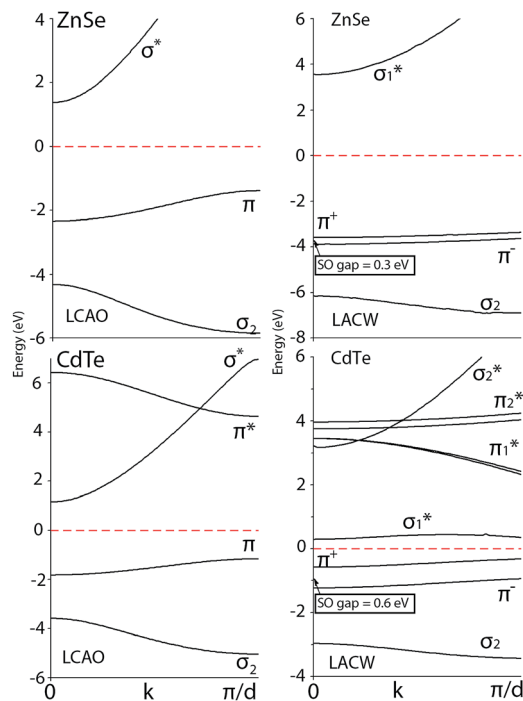


Fig. 4 Non-relativistic LCAO and relativistic LACW band structures of $A^{II}B^{VI}$ chains.

the chains of the elements of the fifth period. The spin-orbit splitting of the non-degenerated σ orbital bands is absent in all the chains.

5. Concluding remarks

Using first principle non-relativistic and relativistic methods, the band structures of the covalent and partially ionic $A^N B^{8-N}$ single-atom-width-chains are calculated. There is a great difference between the relativistic and non-relativistic band structures. The relativistic calculations demonstrate that the spin and orbital motions of electrons are coupled, thereby splitting the π bands. The mass-velocity corrections result in the lowering of all of the valence band levels. The Darwin corrections are several times smaller in comparison to the mass-velocity contributions. The transition from the covalent chains to the partially ionic ones is accompanied by a drastic change in the band structure, due to the presence of the anti-symmetric components of the electron potential in the last case. The spin-orbit interaction complicates the band structure and density of states in the Fermi level region. The transition from the $A^{III}B^V$ chains to $A^{II}B^{VI}$ ones is accompanied by a sharp increase in the band gap. The calculations indicate the metal type of the InSe atomic chain while the CdTe one is an insulator. The calculation results indicate that in the atomic $A^N B^{8-N}$ wires, there are compounds with different electrical properties, from metals to semiconductors and insulators. In the covalent and partially ionic atomic $A^N B^{8-N}$ wires, it is possible to control the electron spin transfer using an external electric field or optical radiation.

6. Appendix: matrix elements of the relativistic terms

The H_{m-v} and H_{Dir} parts of eqn (6) are completely determined by the matrix elements between the eigenfunctions of the non-relativistic Hamiltonian. Following the description given in ref. 46, note that the Hermitian character of the operator p^4 requires that

$$\begin{aligned} & \langle \Psi_{n_2,k}^0(\mathbf{r}) | H_{m-v} | \Psi_{n_1,k}^0(\mathbf{r}) \rangle = -\frac{1}{c^2} \langle \Psi_{n_2,k}^0(\mathbf{r}) | p^4 | \Psi_{n_1,k}^0(\mathbf{r}) \rangle \\ & = -\frac{1}{c^2} \langle p^2 \Psi_{n_2,k}^0(\mathbf{r}) | p^2 \Psi_{n_1,k}^0(\mathbf{r}) \rangle = -\frac{1}{c^2} \langle [H_0 - V(\mathbf{r})] \Psi_{n_2,k}^0(\mathbf{r}) | \\ & \quad \times [H_0 - V(\mathbf{r})] \Psi_{n_1,k}^0(\mathbf{r}) \rangle \\ & = -\frac{1}{c^2} \langle [E_{n_2,k}^0 - V(\mathbf{r})] \Psi_{n_2,k}^0(\mathbf{r}) | \\ & \quad \times [E_{n_1,k}^0 - V(\mathbf{r})] \Psi_{n_1,k}^0(\mathbf{r}) \rangle \end{aligned} \quad (A1)$$

Because of the approximate potential $V(\mathbf{r}) = 0$ in the region outside the MT-spheres, taking eqn (5) into account, we have

$$\begin{aligned} & \langle \Psi_{n_2,k}^0(\mathbf{r}) | H_{m-v} | \Psi_{n_1,k}^0(\mathbf{r}) \rangle = -\frac{1}{c^2} \{ E_{n_2,k} E_{n_1,k} \delta_{n_2,n_1} \\ & \quad + \sum_{\alpha_{MT}} \int_{\Omega_{\alpha_{MT}}} F_{n_2,n_1,k}^{\alpha_{MT}}(\mathbf{r}) \bar{\Psi}_{n_2,k}^0(\mathbf{r}) \Psi_{n_1,k}^0(\mathbf{r}) d\mathbf{r} \} \\ & = -\frac{1}{c^2} \{ E_{n_2,k} E_{n_1,k} \delta_{n_2,n_1} + \sum_{P_2 M_2 N_2} \sum_{P_1 M_1 N_1} \bar{a}_{P_2 M_2 N_2}^{kn_2} a_{P_1 M_1 N_1}^{kn_1} \\ & \quad \times \sum_{\alpha_{MT}} \int_{\Omega_{\alpha_{MT}}} F_{n_2,n_1,k}^{\alpha_{MT}}(\mathbf{r}) \bar{\Psi}_{\alpha_{MT}}^{P_2 M_2 N_2,k}(\mathbf{r}) \Psi_{\alpha_{MT}}^{P_1 M_1 N_1,k}(\mathbf{r}) d\mathbf{r} \} \end{aligned} \quad (A2)$$

where,

$$F_{n_2,n_1,k}^{\alpha_{MT}}(\mathbf{r}) \equiv [E_{n_2,k} - V_{\alpha_{MT}}(\mathbf{r})][E_{n_1,k} - V_{\alpha_{MT}}(\mathbf{r})] - E_{n_2,k} E_{n_1,k} \quad (A3)$$

Finally, taking the analytical form of function $\Psi_{\alpha_{MT}}^{PMN}$ given elsewhere into account,^{8,41-43} we obtain:

$$\begin{aligned} & \langle \Psi_{n_2,k}^0(\mathbf{r}) | H_{m-v} | \Psi_{n_1,k}^0(\mathbf{r}) \rangle = -\frac{1}{c^2} \left\{ E_{n_2,k} E_{n_1,k} \delta_{n_2,n_1} + \frac{\pi}{Q} \sum_{P_2 M_2 N_2} \right. \\ & \quad \times \sum_{P_1 M_1 N_1} \delta_{M_2, M_1} \bar{a}_{P_2 M_1 N_2}^{kn_2} a_{P_1 M_1 N_1}^{kn_1} \\ & \quad \times \{ J_{l_{M_1}}(\kappa_{|M_1|, N_2} a) J_{l_{M_1}}(\kappa_{|M_1|, N_1} a) \}^{-1} \sum_{\alpha_{MT}} r_{\alpha_{MT}}^4 \exp[i(k_{p_1} - k_{p_2}) Z_{\alpha_{MT}}] \\ & \quad \times \sum_{l=|M_1|}^{\infty} (2l+1) \frac{(l-|M_1|)!}{(l+|M_1|)!} \\ & \quad \times \left\{ \mu_{n_2,n_1,k,l,\alpha_{MT}} \bar{a}_{l_{M_1},\alpha_{MT}}^{P_2 M_1 N_2,k} a_{l_{M_1},\alpha_{MT}}^{P_1 M_1 N_1,k} + \dot{\mu}_{n_2,n_1,k,l,\alpha_{MT}} \bar{b}_{l_{M_1},\alpha_{MT}}^{P_2 M_1 N_2,k} b_{l_{M_1},\alpha_{MT}}^{P_1 M_1 N_1,k} \right. \\ & \quad \left. + \dot{\mu}_{n_2,n_1,k,l,\alpha_{MT}} \left[a_{l_{M_1},\alpha_{MT}}^{P_2 M_1 N_2,k} b_{l_{M_1},\alpha_{MT}}^{P_1 M_1 N_1,k} + b_{l_{M_1},\alpha_{MT}}^{P_2 M_1 N_2,k} a_{l_{M_1},\alpha_{MT}}^{P_1 M_1 N_1,k} \right] \right\} \end{aligned} \quad (A4)$$

Here,

$$\mu_{n_2,n_1,k,l,\alpha_{MT}} = \int_0^{r_{\alpha_{MT}}^{MT}} F_{n_2,n_1,k}^{\alpha_{MT}}(\mathbf{r}) u_{l,\alpha_{MT}}^2(\mathbf{r}) r^2 d\mathbf{r}, \quad (A5)$$

$$\dot{\mu}_{n_2,n_1,k,l,\alpha_{MT}} = \int_0^{r_{\alpha_{MT}}^{MT}} F_{n_2,n_1,k}^{\alpha_{MT}}(\mathbf{r}) u_{l,\alpha_{MT}}(\mathbf{r}) \dot{u}_{l,\alpha_{MT}}(\mathbf{r}) r^2 d\mathbf{r}, \quad (A6)$$

$$\ddot{\mu}_{n_2,n_1,k,l,\alpha_{MT}} = \int_0^{r_{\alpha_{MT}}^{MT}} F_{n_2,n_1,k}^{\alpha_{MT}}(\mathbf{r}) \dot{u}_{l,\alpha_{MT}}^2(\mathbf{r}) r^2 d\mathbf{r}, \quad (A7)$$

The Darwin term may be rewritten in a more convenient form by using the following relations⁴⁶

$$\int \nabla \cdot [\bar{\Phi} \Phi'(\nabla V)] d\mathbf{v} = \int_{\text{encl.surf.}} (\bar{\Phi} \Phi' \nabla V) d\sigma \quad (A8)$$

$$\nabla \cdot [\bar{\Phi} \Phi'(\nabla V)] = \bar{\Phi} \Phi'(\nabla^2 V) + (\nabla V) \times (\nabla \bar{\Phi}) \Phi' + (\nabla V) \times (\nabla \Phi') \bar{\Phi} \quad (A9)$$

where the integrals are taken over the volume defined by the periodic boundary conditions. Combining these equations leads to the result, with

$$\Phi = \Psi_{n_2,k}^0(\mathbf{r}) \text{ and } \Phi' = \Psi_{n_1,k}^0(\mathbf{r}):$$

$$\begin{aligned} & \frac{1}{2c^2} \langle \Psi_{n_2,k}^0(\mathbf{r}) | \nabla^2 V(\mathbf{r}) | \Psi_{n_1,k}^0(\mathbf{r}) \rangle \\ & = -\frac{1}{2c^2} \sum_{P_2 M_2 N_2} \sum_{P_1 M_1 N_1} \bar{a}_{P_2 M_2 N_2}^{kn_2} a_{P_1 M_1 N_1}^{kn_1} \\ & \quad \times \sum_{\alpha_{MT}} \int_{\Omega_{\alpha_{MT}}} \left\{ \bar{\Psi}_{\alpha_{MT}}^{P_2 M_2 N_2,k}(\mathbf{r}) [\nabla V_{\alpha_{MT}}(\mathbf{r})] [\nabla \Psi_{\alpha_{MT}}^{P_1 M_1 N_1,k}(\mathbf{r})] \right. \\ & \quad \left. + [\nabla \bar{\Psi}_{\alpha_{MT}}^{P_2 M_2 N_2,k}(\mathbf{r})] [\nabla V_{\alpha_{MT}}(\mathbf{r})] \Psi_{\alpha_{MT}}^{P_1 M_1 N_1,k}(\mathbf{r}) \right\} d\mathbf{r}. \end{aligned} \quad (A10)$$

Taking the analytical form of function $\Psi_{\alpha_{MT}}^{PMN}$ into account, we have

$$\begin{aligned} & \langle \Psi_{n_2,k}^0(\mathbf{r}) | H_{Dir} | \Psi_{n_1,k}^0(\mathbf{r}) \rangle = \\ & \quad -\frac{1}{2c^2} \frac{\pi}{Q} \sum_{P_2 M_2 N_2} \sum_{P_1 M_1 N_1} \delta_{M_2, M_1} \bar{a}_{P_2 M_1 N_2}^{kn_2} a_{P_1 M_1 N_1}^{kn_1} \\ & \quad \{ |J_{l_{M_2}}(\kappa_{|M_2|, N_2} a) J_{l_{M_1}}(\kappa_{|M_1|, N_1} a)| \}^{-1} \\ & \quad \times \sum_{\alpha_{MT}} r_{\alpha_{MT}}^4 \exp\{i[(k_{p_1} - k_{p_2}) Z_{\alpha_{MT}}]\} \\ & \quad \times \sum_{l=|M_1|}^{\infty} (2l+1) \frac{(l-|M_1|)!}{(l+|M_1|)!} \left\{ 2\varsigma_{l,\alpha_{MT}}(u, u') \bar{a}_{l_{M_1},\alpha_{MT}}^{P_2 M_1 N_2,k} a_{l_{M_1},\alpha_{MT}}^{P_1 M_1 N_1,k} \right. \\ & \quad \left. + 2\varsigma_{l,\alpha_{MT}}(\dot{u}, \dot{u}') \bar{b}_{l_{M_1},\alpha_{MT}}^{P_2 M_1 N_2,k} b_{l_{M_1},\alpha_{MT}}^{P_1 M_1 N_1,k} + [\varsigma_{l,\alpha_{MT}}(u, \dot{u}') + \varsigma_{l,\alpha_{MT}}(\dot{u}, u')] \right. \\ & \quad \left. \left[\bar{a}_{l_{M_1},\alpha_{MT}}^{P_2 M_1 N_2,k} b_{l_{M_1},\alpha_{MT}}^{P_1 M_1 N_1,k} + \bar{b}_{l_{M_1},\alpha_{MT}}^{P_2 M_1 N_2,k} a_{l_{M_1},\alpha_{MT}}^{P_1 M_1 N_1,k} \right] \right\}. \end{aligned} \quad (A11)$$

Here,

$$\varsigma_{l,\alpha_{MT}}(u, u') = \int_0^{r_{\alpha_{MT}}^{MT}} u_{l,\alpha_{MT}}(r, E_{l,\alpha_{MT}}) \frac{dV_{\alpha_{MT}}(r)}{dr} u'_{l,\alpha_{MT}}(r, E_{l,\alpha_{MT}}) r^2 d\mathbf{r}, \quad (A12)$$

$$\varsigma_{l,\alpha\text{MT}}(\dot{u}, \dot{u}') = \int_0^{r_{\alpha\text{MT}}} \dot{u}_{l,\alpha\text{MT}}(r, E_{l,\alpha\text{MT}}) \frac{dV_{\alpha\text{MT}}(r)}{dr} \dot{u}'_{l,\alpha\text{MT}}(r, E_{l,\alpha\text{MT}}) r^2 dr, \quad (\text{A13})$$

$$\varsigma_{l,\alpha\text{MT}}(u, u') = \int_0^{r_{\alpha\text{MT}}} u_{l,\alpha\text{MT}}(r, E_{l,\alpha\text{MT}}) \frac{dV_{\alpha\text{MT}}(r)}{dr} u'_{l,\alpha\text{MT}}(r, E_{l,\alpha\text{MT}}) r^2 dr, \quad (\text{A14})$$

$$\varsigma_{l,\alpha\text{MT}}(\dot{u}, u') = \int_0^{r_{\alpha\text{MT}}} \dot{u}_{l,\alpha\text{MT}}(r, E_{l,\alpha\text{MT}}) \frac{dV_{\alpha\text{MT}}(r)}{dr} u'_{l,\alpha\text{MT}}(r, E_{l,\alpha\text{MT}}) r^2 dr. \quad (\text{A15})$$

We will not explain the meaning of the other symbols used in eqn (A3)–(A7) and (A11)–(A15) since they are the same as those described in ref. 8 and 41–43.

The only part of the relativistic matrix element that must be directly calculated between the spinor functions is that arising from H_{S-O} . For this particular case of a linear atomic chain, the matrix elements $\langle \Psi_{n_2, k}^0(\mathbf{r}, \chi_2) | H_{S-O} | \Psi_{n_1, k}^0(\mathbf{r}, \chi_1) \rangle$ have been given in our previous publication⁸ and are not repeated here.

Conflict of interest

The authors declare no competing financial interest.

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